

⑨ 日本国特許庁 (JP)
 ⑩ 公開特許公報 (A)

⑪ 特許出願公開
 昭55-124763

⑤Int. Cl.³
 C 07 D 213/64
 213/70

識別記号
 庁内整理番号
 7138-4C
 7138-4C

⑥公開 昭和55年(1980)9月26日
 発明の数 1
 審査請求 未請求

(全 3 頁)

⑦5-トリフルオロメチル-2-ピリドン誘導体
 ⑧特 願 昭54-32068
 ⑨出 願 昭54(1979)3月19日
 ⑩發 明 者 西山隆三
 高槻市真上町5丁目41番22号
 ⑪發 明 者 藤川敢一
 守山市浮気町321番地の31
 ⑫發 明 者 橋道歟
 草津市野村町221番地

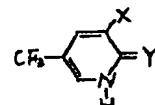
⑦發 明 者 芳賀隆弘
 草津市野村町221番地
 ⑧發 明 者 長谷邦昭
 守山市浮気町321番地の31
 ⑨發 明 者 林弘仁
 守山市浮気町321番地の31
 ⑩出 願 人 石原産業株式会社
 大阪市西区江戸堀1丁目3番11
 号

明細書

1. 発明の名称 5-トリフルオロメチル-2-ピリドン誘導体

2. 特許請求の範囲

1. 一般式

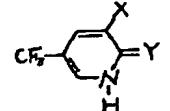


(式中Xは水素原子又はハロゲン原子であり、Yは酸素原子又はイオウ原子である。但し、Xが水素原子の場合、Yはイオウ原子である。)で表わされる5-トリフルオロメチル-2-ピリドン誘導体。

3. 発明の詳細な説明

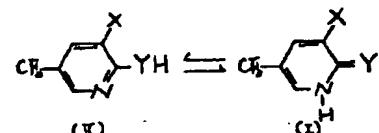
本発明は医薬、農薬、染料などの中间原料として有用で、新規な5-トリフルオロメチル-2-ピリドン誘導体に関する。

詳しくは本発明は一般式



(式中Xは水素原子又はハロゲン原子であり、Yは酸素原子又はイオウ原子である。但し、Xが水素原子の場合、Yはイオウ原子である。)で表わされる5-トリフルオロメチル-2-ピリドン誘導体である。

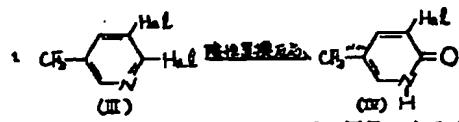
前記一般式(I)の5-トリフルオロメチル-2-ピリドン誘導体は、次に示すような互変異性として存在することができる。



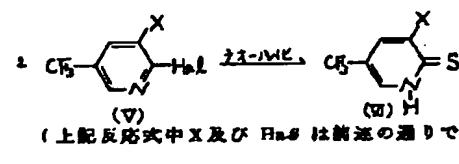
(式中X及びYは前述の通りである)

前記一般式(I)において、Xで表わされるハロゲン原子としては弗素、氯素、溴素、沃素が挙げられる。

本発明の5-トリフルオロメチル-2-ビリ
ン衍生物は通常、例えば下記方法によって製
造される。



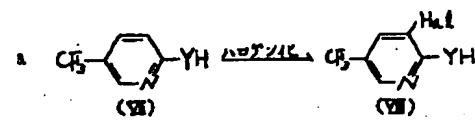
一般に上記反応はジメチルスルホキシド、ジ
メチルホルムアミドなどの非プロトン性極性溶
媒中、水酸化ナトリウム、水酸化カリウムなど
のフルカリ水溶液を用いて $50 \sim 150^\circ\text{C}$ 、0.1
~ 10時間で行なわれる。



一般に上記反応はメタノール、エタノールな
どのアルコール類、ジメチルスルホキシド、ジ
メチルホルムアミドなどの非プロトン性極性溶

- 3 -

特開昭55-124763(2)
媒などの溶媒中、チオ尿素、硫化ソーダ、チオ
硫酸ソーダ、N,N-ジメチルジチオカルバミ
ン酸ソーダなどのチオール化剤を用いて $50 \sim$
還流温度 $0.5 \sim 10$ 時間で行なわれる。



一般に上記反応は四塩化炭素、クロロホルム、
酢酸、二硫化炭素、水、非プロトン性極性溶媒
などの溶媒中、塩素ガス、臭素、チオニルクロ
ライド、スルフィルクロライドなどのハロゲン
化剤を用いて $0 \sim 100^\circ\text{C}$ 、0.5~10時間で
行なわれる。

本発明化合物は、例えばハロゲン化ニトロベ
ンゼン類と結合させて4-(5-トリフルオロメ
チルピリジン-2-イルオキシ)ニトロベン
ゼン類を生成させ、これを還元して得られる4
-(5-トリフルオロメチルピリジン-2-イ
ルオキシ)アニリン類とベンゾイルイソシアニ

- 4 -



ト類とを反応させることによりN-ベンゾイ
ル-N'-(4-(5-トリフルオロメチルピリ
ジン-2-イルオキシ)フェニル)ウレア系化
合物に誘導できる。詳しくは本発明化合物の3
-クロロ-5-トリフルオロメチル-2-ビリ
ドンと3,4,5-トリクロロニトロベンゼンとを
結合、還元して3,5-ジクロロ-4-(3-ク
ロロ-5-トリフルオロメチルピリジン-2-
イルオキシ)アニリンを得、更にこのものと2
,6-ジフルオロベンゾイルイソシアネートとを
反応させると、N-(2,6-ジフルオロベンゾ
イル)-N'-(3,5-ジクロロ-4-(3-ク
ロロ-5-トリフルオロメチルピリジン-2-
イルオキシ)フェニル)ウレアを得ることがで
きる。このものは殺虫剤の有効成分として優れ
た活性を示し、種々の有害虫、特に有害昆虫の
防除に有効であって、例えばこの化合物 100
ppm水分散液にキャベツの葉片を浸漬し、それ
を風乾してそこへ2~3合のコナガの幼虫を放
ち、8日目に生死を判定した結果、100%の

死虫率が得られた。

次に本発明化合物の具体的合成例を記載する。

合成例 1 3-クロロ-5-トリフルオロメチ
ル-2-ビリドン

〔A〕

5-トリフルオロメチル-2-ビリドン 0.2 g
をクロロホルム 20 ml に溶解させ、 50°C
に加熱して塩素ガスを1時間搅拌下に通じ
た。反応終了後、クロロホルムを留去し、ト
ルエン-二オヘキサンの混合溶媒で再結晶し
て融点 $144 \sim 147^\circ\text{C}$ の目的物 0.15 g を
得た。

〔B〕

水酸化ナトリウム 2.4 g を水 2.5 ml に溶
解させた水溶液に2,3-ジクロロ-5-トリ
フルオロメチルピリジン 4 g を加え、更にジ
メチルスルホキシド 1.25 ml を加えて加熱し、
 110°C で1時間搅拌下に反応させた。反応
終了後生成物を放冷し、酸性にして
沈殿物を、このものを通過して目的 2.5

- 5 -

- 6 -

タを得た。

合成例2 5-トリフルオロメチル-2-テオ
ピリドン

2-クロロ-5-トリフルオロメチルヒリジン4タとテオ尿素167タとをエタノール30mlに溶解させ、加熱して還流状態で3時間攪拌下に反応させた。その後、水酸化カリウム水溶液123タを徐々に加えて還流状態で1時間反応させた。反応終了後、生成物を放油し、希アルカリ水溶液中に投入して塩化メチレンで洗浄し、酢酸で酸性にした。次いで、塩化メチレンで抽出し、抽出層を水洗後無水硫酸ナトリウムで乾燥させ、塩化メチレンを留去して融点147~150℃の目的物2.1タを得た。

合成例3 3-プロモ-5-トリフルオロメチル-2-ピリドン

5-トリフルオロメチル-2-ピリドン0.4タを酢酸10mlに溶解させ、そこへ臭素0.4タを加えて攪拌下で4時間反応させた。反

特許昭55-124763(3)

応終了後、酢酸を留去し、塩化メチレン-ローヘキサンの混合溶媒で再結晶して融点162~165℃の目的物0.45タを得た。

合成例4 3-クロロ-5-トリフルオロメチル-2-テオピリドン

2-クロロ-5-トリフルオロメチルヒリジン4タに代えて2,3-ジクロロ-5-トリフルオロメチルヒリジン4.75タを用いる以外は前記合成例2と同様にして反応を行ない、後処理を行なって融点125~128℃の目的物1.9タを得た。

特許出願人 石原産業株式会社

UNITED STATES PATENT OFFICE

HERMANN THOMS, OF BERLIN, GERMANY.

PROCESS OF MAKING PARA-PHENETOL CARBAMIDE.

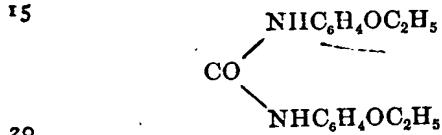
SPECIFICATION forming part of Letters Patent No. 502,504, dated August 1, 1893.

Application filed November 18, 1892. Serial No. 452,446. (Specimens.)

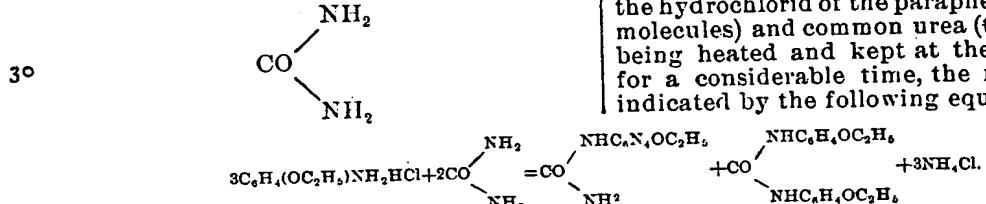
To all whom it may concern:

Be it known that I, HERMANN THOMS, chemist, a subject of the Emperor of Germany, residing in the city of Berlin, German Empire, have invented certain new and useful Improvements in the Production of Para Phenetyl Carbamide; and I do hereby declare that the following is a full, clear, and exact description of the invention, such as will enable others skilled in the art to which it appertains to make and use the same.

My previous researches (published in the *Pharm. Centralhalle*, March 24, 1892,) have shown that di-para-phenetylurea



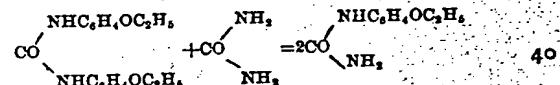
may be readily obtained, in addition to the hydrochlorid of phenetidin, by causing carbonylchlorid to act on a solution of para phenetidin in toluene. Since then I have found 25 that this body, when heated for several hours with common urea



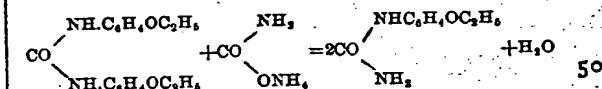
This process will yield, in addition to the para-phenetol carbamide, diparaphenetylurea. The paraphenetolcarbamide crystallizes from the hot filtrate.

5 The paraphenetolcarbamide obtained as described from diparaphenetylurea, or from paraphenetidin by the action of common urea or the carbamide salt of ammonia, or commercial ammonium carbonate, melts at a temperature approaching 170° centigrade, and 10 has a sweet taste of extraordinary intensity which renders it suitable for industrial application as a sweetening substance. According to physiological experiments, the new 15 substance is quite harmless to the human organism.

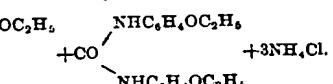
in equimolecular proportions in a closed vessel, and at a temperature ranging between 150° and 160° centigrade, is easily converted 35 into the para phenetol carbamide as indicated by the following equation:



Instead of the common urea the carbamide salt of ammonia or commercial ammonium carbonate may be employed. The reaction takes place in the first case as indicated by 45 the following equation:



I have found also, that instead of the di-paraphenetylurea, paraphenetidin or the hydrochlorid of para-phenetidin may be employed, the latter being either treated in a 55 closed vessel with common urea, or the carbamide salt of ammonia, or with commercial ammonium carbonate at a temperature of 160° centigrade; or an aqueous solution of the hydrochlorid of the paraphenetidin (three 60 molecules) and common urea (two molecules) being heated and kept at the boiling point for a considerable time, the reaction being indicated by the following equation:



Having thus described my invention, what I claim as new therein, and desire to secure by Letters Patent, is—

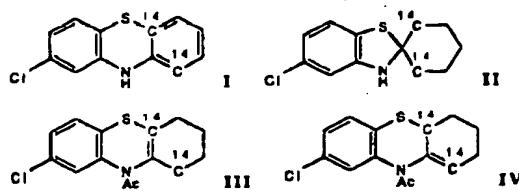
1. The process of obtaining paraphenetol carbamide, by the reaction of a para salt of phenetidin on a substance such as common urea in about the proportions set forth.

2. The process of obtaining para-phenetolcarbamide, which consists in boiling an aqueous solution of para-phenetidin-hydrochlorid with common urea in about the proportions set forth.

HERMANN THOMS.

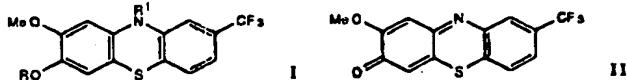
Witnesses:

FRITZ RINDEL,
AUG. FRAHNE.



to give the tetrahydrophenothiazine olefin mixt. III and IV which was directly converted to labeled I via treatment with DDQ in refluxing benzene followed by hydrolysis of the acetyl group.

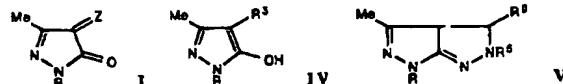
91: 74554j **Synthesis of 7,8-disubstituted metabolites of triflupromazine: 2-(trifluoromethyl)-7,8-dimethoxy-10-[3-(dimethylamino)propyl]-phenothiazine and related compounds.** Mital, R. L.; Mittal, Madhu; Laxmi, V.; Mittal, Suresh; Shukla, A. P. (Dep. Chem., Univ. Rajasthan, Jaipur, 302 004 India). *J. Inst. Chem. (India)* 1978, 50(4), 159-61 (Eng). Phenothiazine I [R = Me, R¹ = (CH₂)₃NMe₂], a



metabolite of triflupromazine was prep'd. Thus, condensation of 2,4-H₂N(F₃C)C₆H₃SH Zn salt with 2-chloro-5-methoxy-p-benzoquinone in refluxing EtOH 4 h gave II quant., II was reduced with Na₂S₂O₄ in aq. Me₂CO to give 90% phenothiazinol I (R = R¹ = H). The product was O-methylated with Me₂SO₄ in Me₂CO contg. Na₂S₂O₄ and aq. KOH 4 h at 60° and the product ether I (R = Me, R¹ = H) (67% yield) was N-alkylated by Cl(CH₂)₃NMe₂ in Me₂SO contg. NaH 2 h at room temp. to give I [R = Me, R¹ = (CH₂)₃NMe₂], characterized as its maleate.

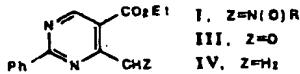
DIAZINES

91: 74555k **Reactions of 3-methyl-1-aryl-Δ²-pyrazolin-5-ones with aromatic aldehydes, aryl diazonium chlorides and of their products 3-methyl-1-aryl-4-arylidene-Δ²-pyrazolin-5-ones with secondary amines, hydrazines, dialkyl phosphites, Grignard reagents, ethyl aceto- or cyanoacetate and cyclohexanone.** Zimaity, T.; Afsah, E.; Abbas, M. (Fac. Sci., Mansoura Univ., Mansoura, Egypt). *Indian J. Chem., Sect. B* 1978, 16B(10), 876-9 (Eng). Reactions of I (R = p-ClC₆H₄),



p-O₂NC₆H₄; Z = H₂ (II) with R¹CHO (R¹ = p-MeOC₆H₄, O₂NC₆H₄, Me₂NC₆H₄; thienyl) gave I (Z = CHR₁) (III). II and p-ClC₆H₄N₂Cl gave I (Z = H, R¹ = N:NC₆H₄-Cl-p). Mannich reaction of II gave I (Z = H, R¹ = R² = p-ClC₆H₄, Me). III and piperidine gave IV (R³ = p-MeOC₆H₄CHR⁴, R⁴ = piperidino, etc.). Cyclization of III with N₂H₄ and PhNHNNH₂ gave V (R⁵ = Ph, H; R⁶ = p-MeOC₆H₄, etc.). Reactions of II with dialkyl phosphite, Grignard reagents, Et acetoacetate, NCC₂CO₂Et and cyclohexanone gave compds. related to I and IV.

91: 74556m **Synthesis and biological activity of α-(5-ethoxycarbonyl-2-phenyl-4-pyrimidinyl)-N-substituted nitrones.** Roy, S. K.; Rao, K. Srinivasa; Reddi, G. S.; Sachdeva, Meena (Res. Dev. Dep., Indian Drugs and Pharm. Ltd., Hyderabad, India). *Indian J. Chem., Sect. B* 1978, 16B(10), 907-9 (Eng).

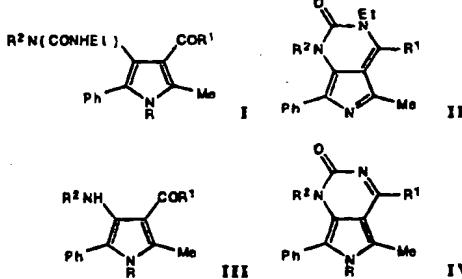


Title compds. I (R = Et, Pr, Bu, CH₂CH₂OH, Ph, PhCH₂, o-MeC₆H₄, p-ClC₆H₄ (II), p-MeSO₂C₆H₄) were prep'd. by treating RNHOH with pyrimidinecarboxaldehyde III, which was prep'd. by Kroehnke oxidn. of IV. I at 25-200 µg/mL were fungicidal against dermatophytes. II killed *Mycobacterium tuberculosis* at 25 µg/mL.

91: 74557n **Pyrimidines. Part LXXXVI. tert-Butylation of quinazoline.** De Bie, D. A.; Nagel, A.; Van der Plas, H. C.; Geurtsen, G.; Koudijs, A. (Lab. Org. Chem., Agric. Univ., Wageningen, Neth.). *Tetrahedron Lett.* 1979, (7), 649-52 (Eng). Quinazoline (I) is present in soln. at pH 3 as its cationic covalent hydrate; and treatment of an aq. soln. of I with excess Me₃CO₂H and ammonium peroxydisulfate, in the presence of a catalytic amt. of AgNO₃ at 40° and at pH 1, gave 2-tert-butyl-3,4-dihydro-4-oxoquinazoline (II), quant. Similar treatment of I at 70° and at pH 5 for 2 h gave a 4:3:2 mixt. of 2-tert-butyl-quinazoline (III), 4-tert-butylquinazoline (IV), and 2,4-di-tert-butylquinazoline (V), whereas similar treatment of I at 70° and at pH 4 gave mainly 2-HCO₂C₆H₄NHCHO and 2-HCO₂C₆H₄NH₂ (VI). At pH 3, VII was the main product together with III, IV,

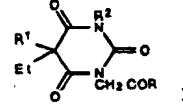
V, and 4-tert-butyl-3,4-dihydroquinazoline. The formation of II, III, IV, V, and VI is discussed.

91: 74558p **Synthesis and antiinflammatory properties of some pyrrolo(1H,3H)[3,4-d]pyrimidin-2-ones and pyrrolo-(1H,6H)[3,4-d]pyrimidin-2-ones.** Tarzia, G.; Panzone, G.; Schiatti, P.; Selva, D. (Dep. Org. Chem., Lepetit Res. Lab., Milan, Italy). *Farmaco, Ed. Sci.* 1979, 34(4), 316-30 (Eng).



The cyclocondensation reaction of pyrroles I (R = H, Me, Et; R¹ = Me, Ph; R² = Et, H, CHMe₂) in MeOH contg. HCl yielded pyrrolopyrimidinones II, and III (R, R¹, and R² same as above), which reacted with NaOON in room temp. to give IV; II and IV exhibited antiinflammatory activity. III (R = R¹ = Et, R² = Me) in HOAc was added to NaOON in H₂O, and the mixt. was kept 4 h at room temp. to give IV (R = R¹ = Et, R² = Me).

91: 74559q **Synthesis and pharmacological screening of some N-carboxymethylbarbituric acid derivatives.** I. Mirek, Julian; Adamczyk, Maciej; Chojnacka-Wojcik, Ewa; Naparzewska, Anna (Inst. Chem., Jagellonian Univ., 30-060 Krakow, Pol.). *Pol. J. Pharmacol. Pharm.* 1978, 30(5), 685-93 (Eng). Methylphenobarbital or barbital were N-alkylated with



ClCH₂CO₂Me or BrCH₂CO₂Et in PhMe contg. K₂CO₃ to give 87-90% carbalkoxy derivs. I (R = OMe, OEt, R¹ = Ph, R² = Me) or 85-6% I (R = OMe, R² = CH₂CO₂Me, R = OEt, R¹ = CH₂CO₂Et, R¹ = Et). Hydrolysis of these esters with refluxing concd. HCl gave 90% I (R = OH, R¹ = Ph, R² = Me) or 95% I (R = OH, R¹ = Et, R² = CH₂CO₂H) which were converted into 95% the corresponding acid chlorides with SOCl₂. I (R = Cl, R¹ = Ph, R² = Me) was treated with 2 mol-equiv amines to give 82-90% amides I (R = 2-, 4-HOCC₆H₄NH, 3-pyridylamino, 4-pyridylmethylamino). I (R = Cl, R¹ = Et, R² = CH₂COCl) was treated with 4 mol-equiv amines to give 89-92% diamides I (R = 2-, 4-HOCC₆H₄NH, 3-pyridylamino, 4-pyridylmethylamino, morpholino; R¹ = RCOCH₂). The amides had no anticonvulsant activity and showed only slight sedative and analgesic action.

91: 74560h **Photolysis of thiopyrimidine derivatives. Part II. 2-(Methylthio)-6-methyluracil and 2-(methylthi)-6-ethyluracil.** Golankiewicz, Krzysztof; Szajda, Maria; Wyrzykiewicz, Elzbieta (Inst. Chem., A. Mickiewicz Univ., 60780 Poznan, Pol.). *Pol. J. Chem.* 1979, 53(2), 529-31 (Eng). Irradn. of I (R = Me, Et) in Me₂CO at λ > 254 nm gives 20.5% II (R = Me, Et). The hydrolysis of II (R = Me) gave III which on irradn. (in acidic, basic, or neutral H₂O) at 254 nm gave 6-methyluracil; this established the anti-configuration for II (R = Me). The photodimerization of I (R = alkyl) was contrasted to the lack of photodimerization of I (R = CO₂H).

91: 74561j **Succinat dehydrogenase inhibitory activity of new 1-aryl-3-(N,N-dimethylaminopropyl) thiobarbiturates.** Tripathi, Shephali; Pandey, B. R.; Raman, K.; Barthwal, J. P.; Kishor, K.; Bhargava, K. P. (King Georg's Med. Coll., Lucknow Univ., Lucknow, India). *Eur. J. Med. Chem. - Chim. Ther.* 1979, 14(2), 133-4 (Eng). Thiobarbiturates I (R = Ph, isomeric tolyl, xylyl, or anisyl, 2-EtOC₆H₄, 2- or 4-ClC₆H₄, 4-BrC₆H₄) were prep'd. by treating Me₂NCH₂CH₂CH₂NH₂ with RNCS and cyclocondensing product thioureas Me₂NCH₂CH₂CH₂NHC(S)NHR with malonic acid. I inhibited (15.1-75.50%) succinate dehydrogenase in vitro activity of rat brain homogenate.

